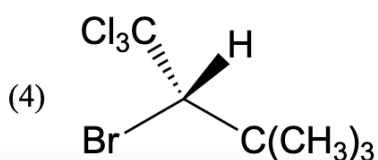
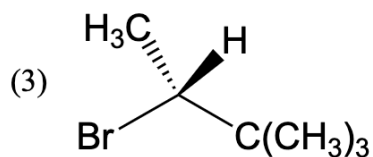
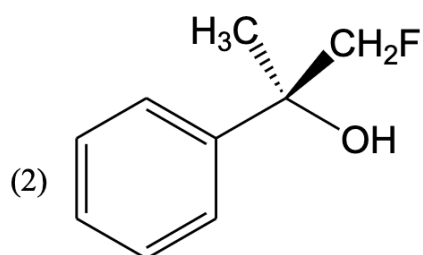
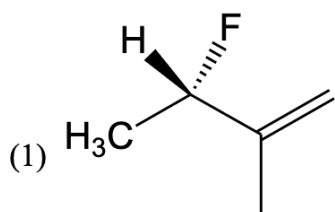
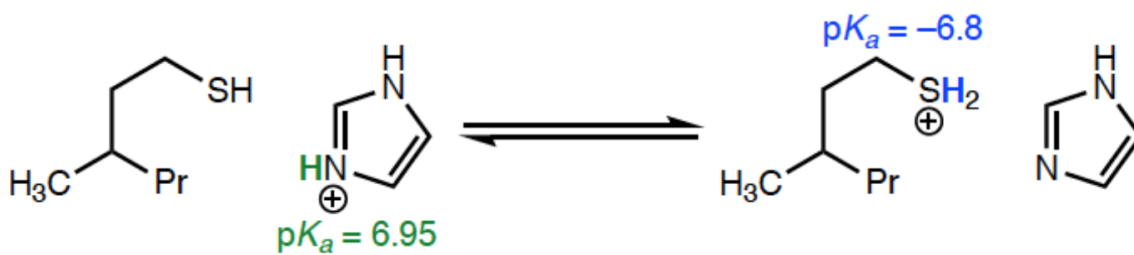


1. Assign R/S centers.



2.

Identify whether the acid-base equilibrium lies to the **left** or to the **right** in the reactions below. (2 pts each)



left

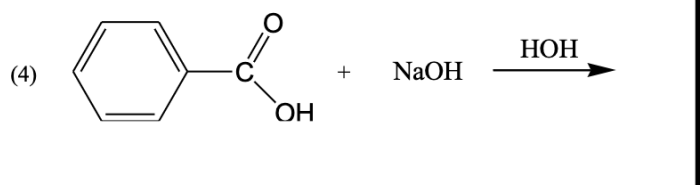
right

3. What is the index of hydrogen deficiency of a compound with a molecular formula of  $C_9H_{11}NO$ ?

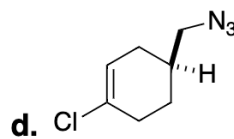
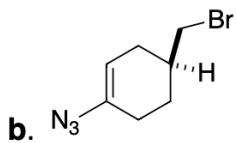
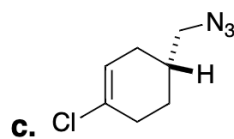
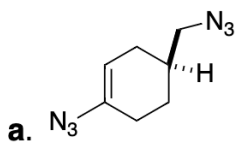
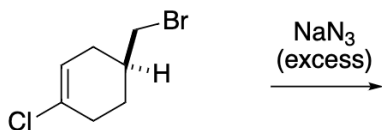
- (a) 3 (b) 4 (c) 5 (d) 6



Complete the following acid-base reaction (write the products).



5. Azide anion is a very good nucleophile. Predict the major product from the following reaction?



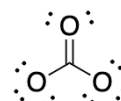
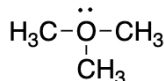
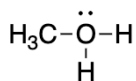
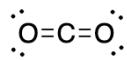
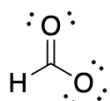
6.

Which of the following would react most quickly in an S<sub>N</sub>1 reaction with acetic acid?

- a. methyl fluoride
- b. ethyl chloride
- c. isopropyl chloride
- d. tert-butyl bromide



7. Which structure(s) contain(s) an oxygen that bears a **formal charge of +1**?



I

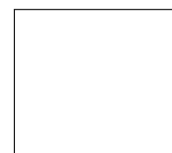
II

III

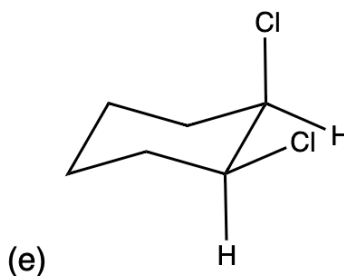
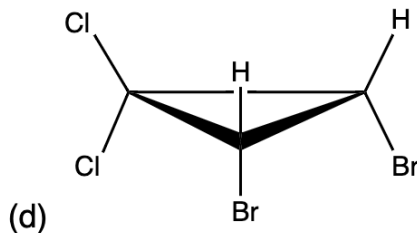
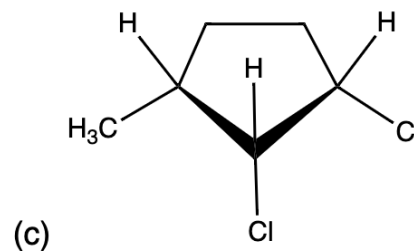
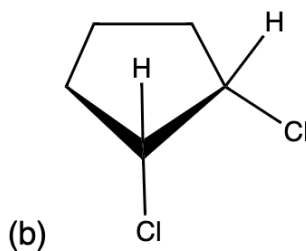
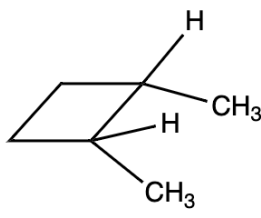
IV

V

- (a) I and II
- (b) III and IV
- (c) V
- (d) III
- (e) I and V

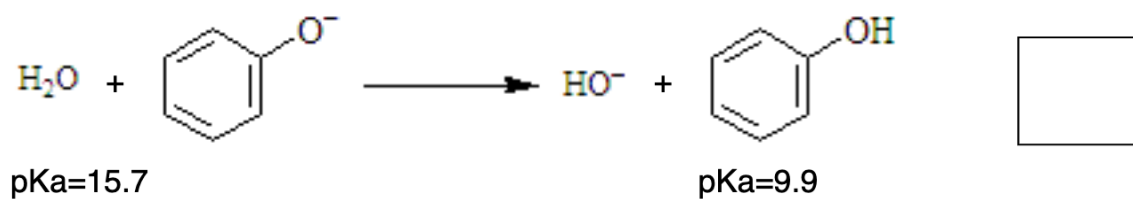


10. Which compounds is **not** a meso compound?

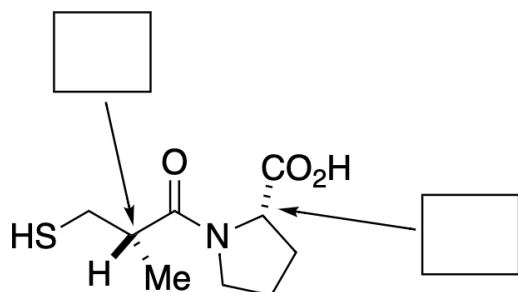
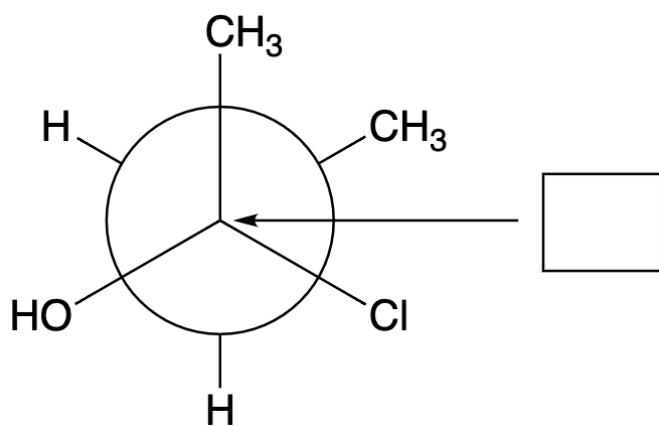


True or False:

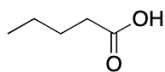
12. The given reaction should be favorable based on the pKa values shown.



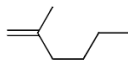
12. For each of the indicated chiral centers below, provide the correct R or S designation. (6 pts)



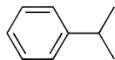
12. Match each compound to its IR spectrum. (12 pts)



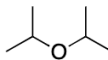
**A**



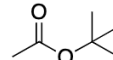
**B**



**C**



**D**

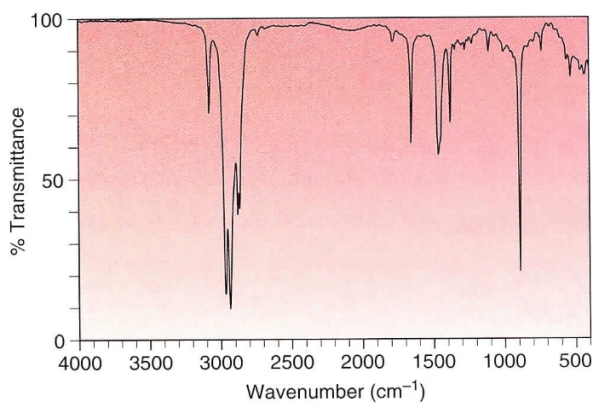


**E**

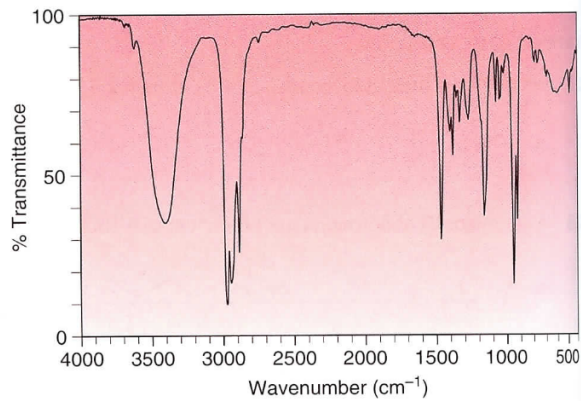


**F**

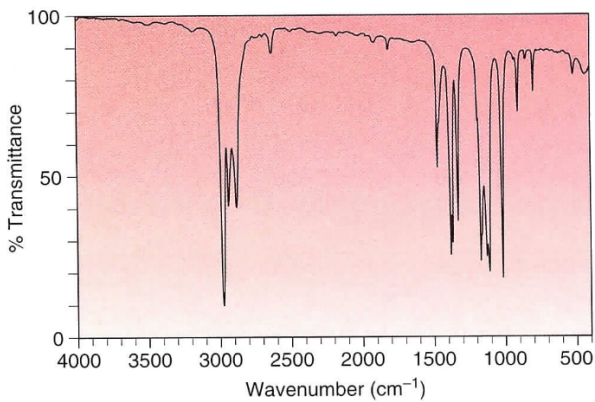
Spectrum [1]



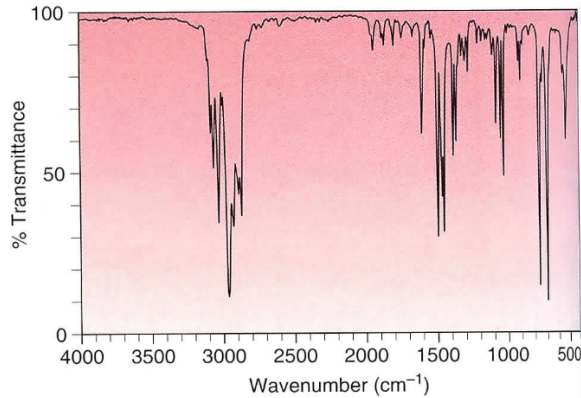
Spectrum [2]



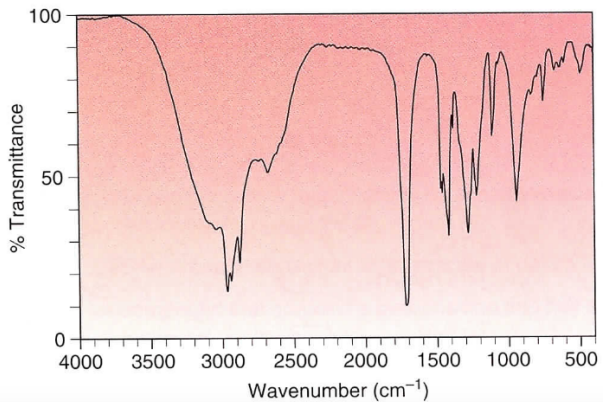
Spectrum [3]



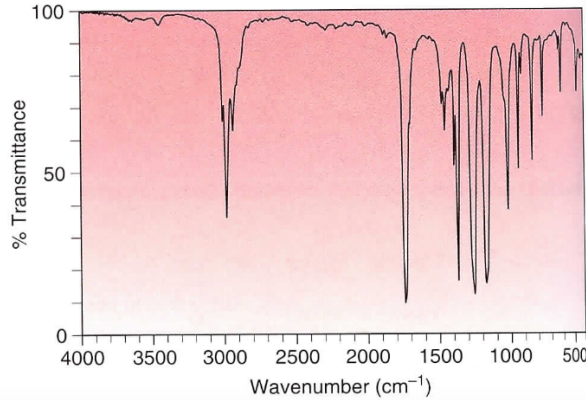
Spectrum [4]



Spectrum [5]



Spectrum [6]

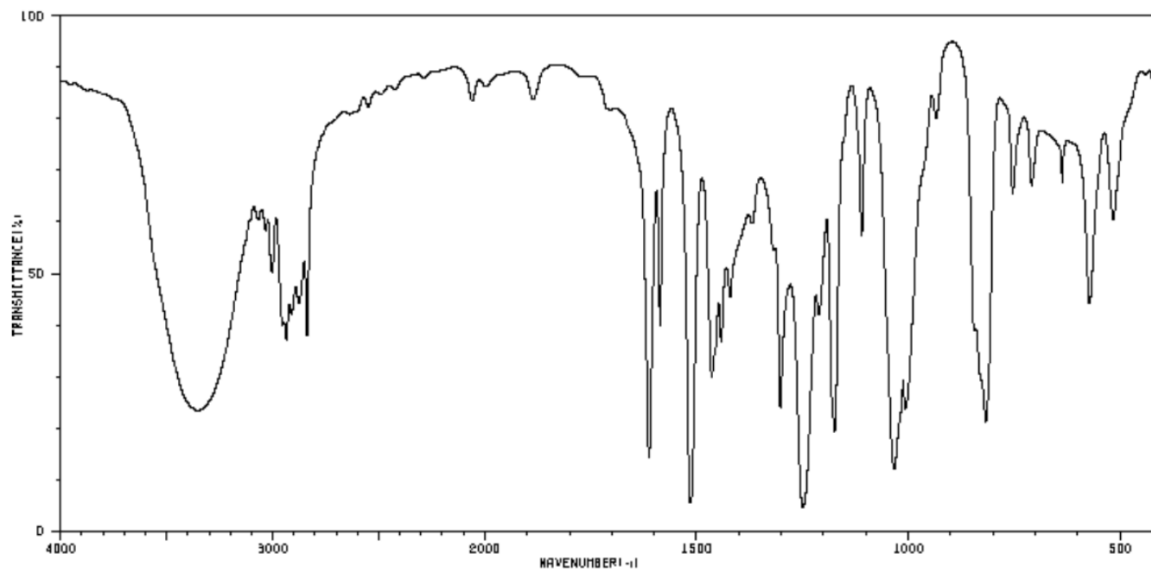


14. For the problem shown below, answer the questions and draw the structure that corresponds to the following spectra.

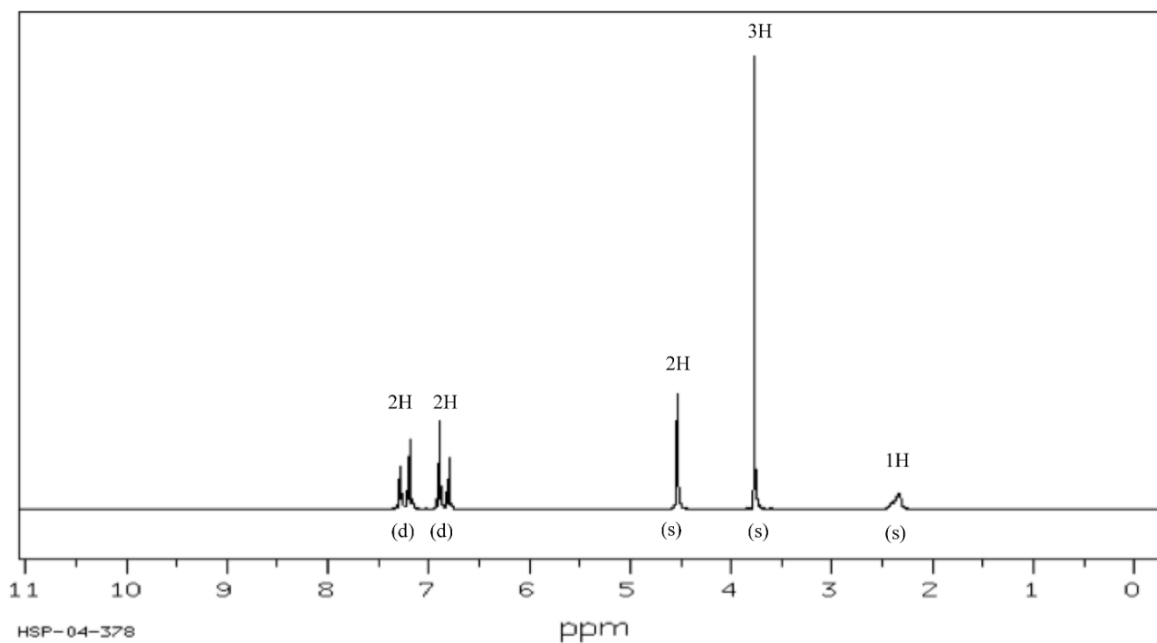
Empirical Formula:  $C_4H_5O$

Mass Spec:  $M^+ m/e = 138$

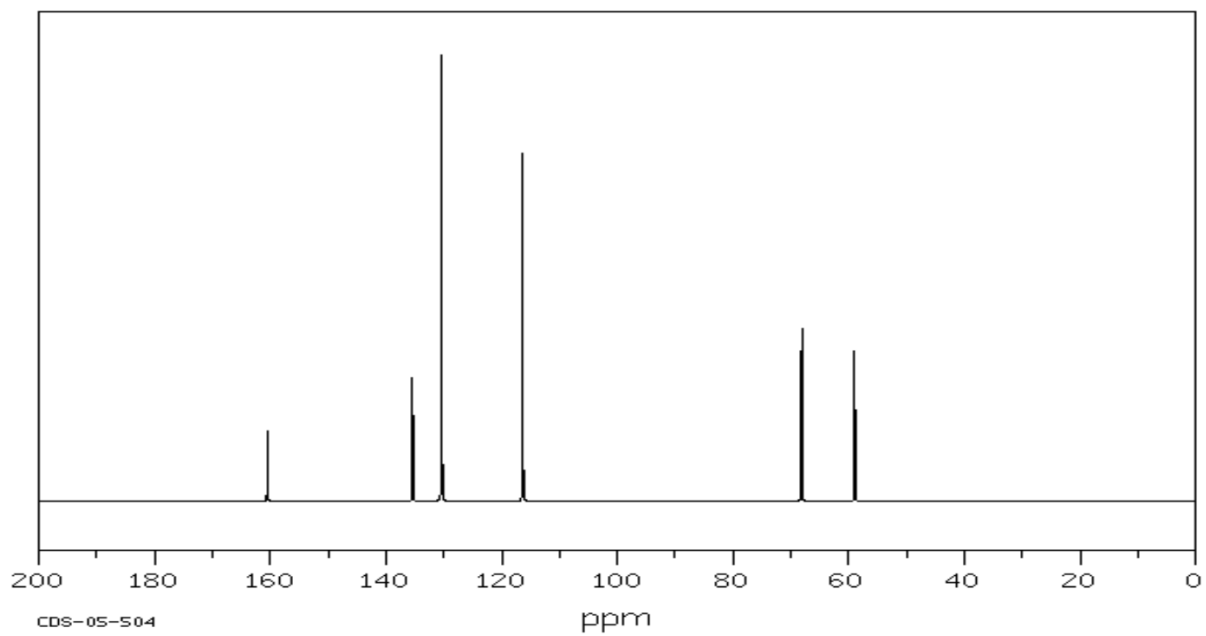
IR



$^1H$  NMR



<sup>13</sup>C NMR



---

**Determination of the Molecular Formula (3pts)**

- a) What the molecular weight of the unknown compound? \_\_\_\_\_
- b) What is the correct molecular formula of the unknown compounds? \_\_\_\_\_
- c) How many degrees of unsaturation does this compound have? \_\_\_\_\_

**Analysis of the IR spectrum (3 pts)**

d) Which of the following bonds are present?

(circle all that are present)    O-H                    C-O                    C=O

e) Based on the analysis of the IR, what type(s) of functional group(s) is/are present?

\_\_\_\_\_

**Analysis of the <sup>1</sup>H NMR spectrum (5 pts)**

f) How many different protons are there in the molecule? \_\_\_\_\_

g) Which of the following are present? (circle all that apply)

Et    iPr    tBu    H on a heteroatom    vinylic H (i.e. an alkene)    aromatic H

h) How many aromatic hydrogens are there? \_\_\_\_\_

i) How many substituents are there on the aromatic ring? \_\_\_\_\_



**Analysis of the  $^{13}\text{C}$  NMR (4pts)**

j) How many different carbons are there in the molecule? \_\_\_\_\_

k) How many types of aromatic carbons are there in the molecule? \_\_\_\_\_

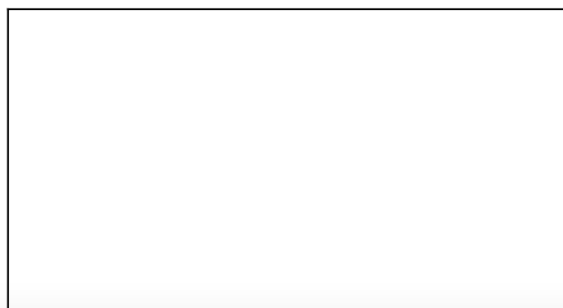
l) Is there a peak arising from a carbonyl in the  $^{13}\text{C}$  NMR spectrum? (circle one) Yes No

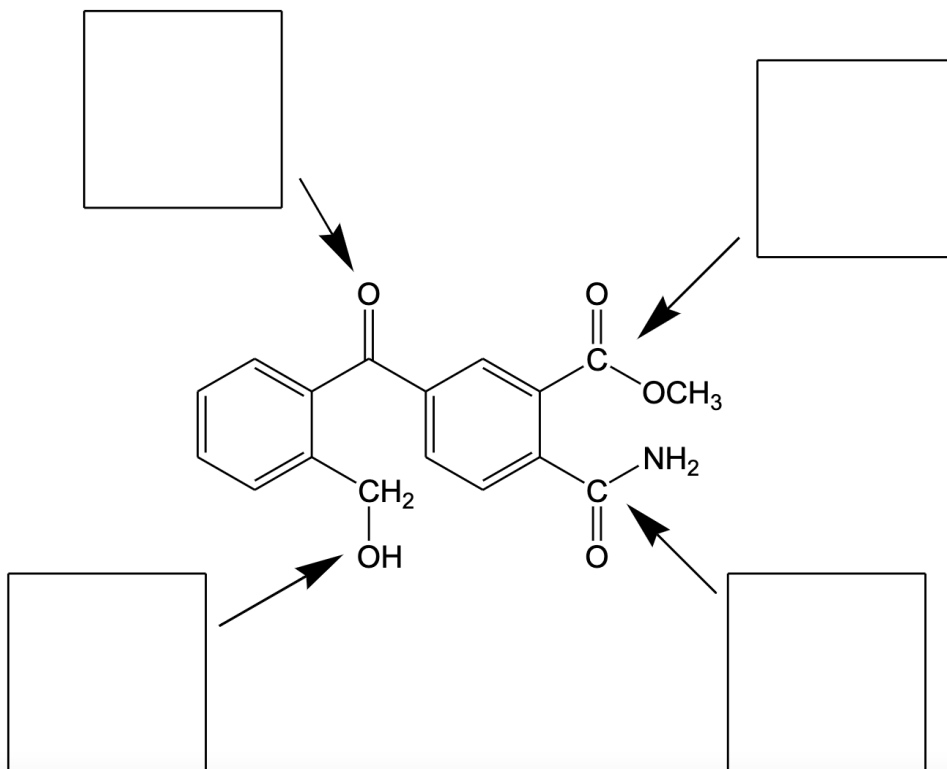
m) How many types of  $\text{sp}^3$  carbons are there? \_\_\_\_\_

**Putting it all together (5 pts)**

n) Suggest a single structure for the molecule that is consistent with all of the data presented.

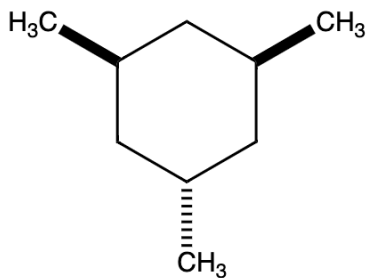
*Note: Most of the credit for this problem comes from answering the questions above in the spaces provided.*

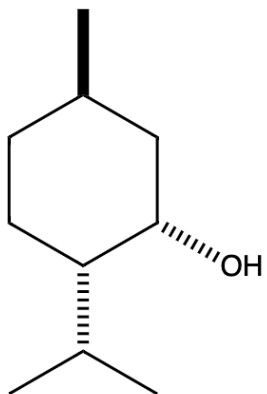




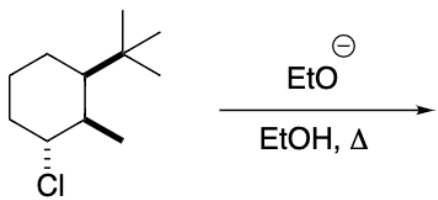
Note this is strictly for helping you on other problems (i.e. IR)! It's important you're familiar with some functional groups. :) You will not be tested about nomenclature only.

16. Draw the most stable chair conformation of the following compounds. (pts)

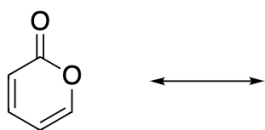




17.



18. Provide **three possible resonance contributors** for the following compound (12 pts total).



19.

